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Synthesis of Phosphorus Containing Polymers by Phase Transfer Catalysis (PTC). I. Comparative Study between Liquid-Liquid and Liquid-Vapor Systems

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Polyphosphonates were synthesized by interfacial alkaline polycondensation of bispherol A (BA) with phenyldichlorophosphonate (PPD) and cyclohexyldichlorophosphonate (CPD) in liquid-liquid $^{\{1\}}$ (I-I) and liquid-vapor $^{\{2\}}$ (I-v) systems.

$$nCl = P - Cl + nHO - CH_3$$

$$CH_3$$

$$PTC$$

Different base concentrations, molar ratios (PPD/BA or CPD/BA) and temperatures were applied and the yield (η) and inherent viscosity (η_{iv}) of the products were determined. In both systems the best results were obtained with 1-2 M aqueous solutions NaOH (i.e. with 1.5 M NaOH, for R=C₆H₅, in 1-1, η =60% and η_{inh} =0.80 dl/g, in 1-v, η =65%, η_{inh} =0.95dl/g; and for R=C₆H₁₁, in 1-1, η =55%, η_{inh} =0.72dl/g; in 1-v, η =60%, η_{inh} =0.90dl/g).

Concerning to the molar ratios (varied in either direction from unity) better results were found with an exces of phosphonic dichloride (with PPD:BA=0.75, in I-I, η =45% and η_{inh} =0.58 dl/g, in I-v, η =48%, η_{inh} =0.62dl/g; with PPD:BA=1.30, in I-I, η =65%, η_{inh} =0.78dl/g; in I-v, η =70%, η_{inh} =0.95dl/g; with CPD:BA=0.75, in I-I, η =40% and η_{inh} =0.57 dl/g, in I-v, η =52%, η_{inh} =0.65dl/g; with CPD:BA=1.30, in I-I, η =65%, η_{inh} =0.75dl/g; in I-v, η =68%, η_{inh} =0.94dl/g)

The influence of the temperature are as fallows: in I-I, the yields and inherent viscosities increase with decrease of the temperature, whereas, in I-v will show an increase in yield and inherent visosity as the temperature is increased. (i.e., at $T=0^{0}C$, for $R=C_{6}H_{5}$, in I-I, $\eta=72\%$ and $\eta_{inh}=0.85$ dl/g,; and for $R=C_{6}H_{11}$, in I-I, $\eta=70\%$, $\eta_{inh}=0.82$ dl/g; at $T=25^{0}C$, for $R=C_{6}H_{5}$, in I-I, $\eta=40\%$ and $\eta_{inh}=0.48$ dl/g,; and for $R=C_{6}H_{11}$, in I-I, $\eta=32\%$, $\eta_{inh}=0.30$ dl/g; at $T=35^{0}C$, for $R=C_{6}H_{5}$, in I-v, $\eta=30\%$ and $\eta_{inh}=0.42$ dl/g,; and for $R=C_{6}H_{11}$, in I-v, $\eta=75\%$ and $\eta_{inh}=0.72$ dl/g,; and for $R=C_{6}H_{11}$, in I-v, $\eta=70\%$, $\eta_{inh}=0.70$ dl/g).

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